

# Effectiveness of a bio-catalytic agent used in the bioremediation of crude oil-polluted seawater

Glenda Teran-Cuadrado\*, Camilo Ortega-Vega, Henrique Alves-de-Brito, Diego Quiñones-Murillo, Mayerlenis Jiménez-Rojas.

Published on April 22nd, 2021

<https://doi.org/10.1016/j.heliyon.2021.e06926>

# OUTLINE



1

*Introduction*

2

*Materials and Methods*

3

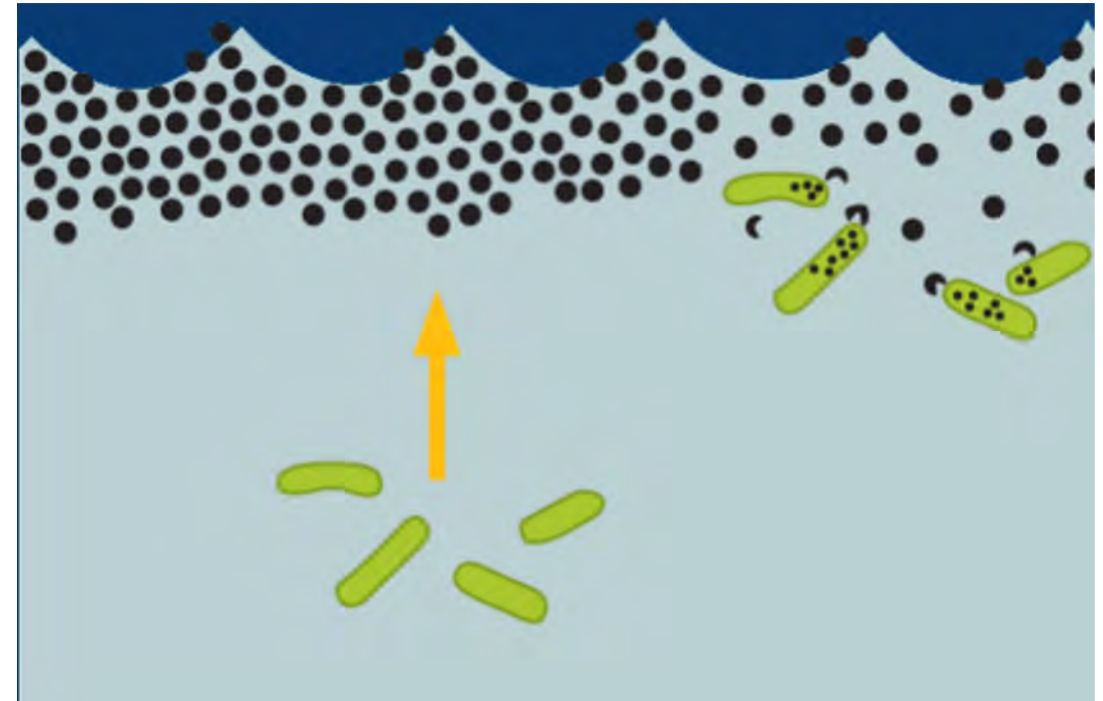
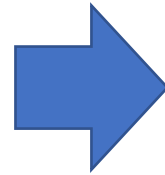
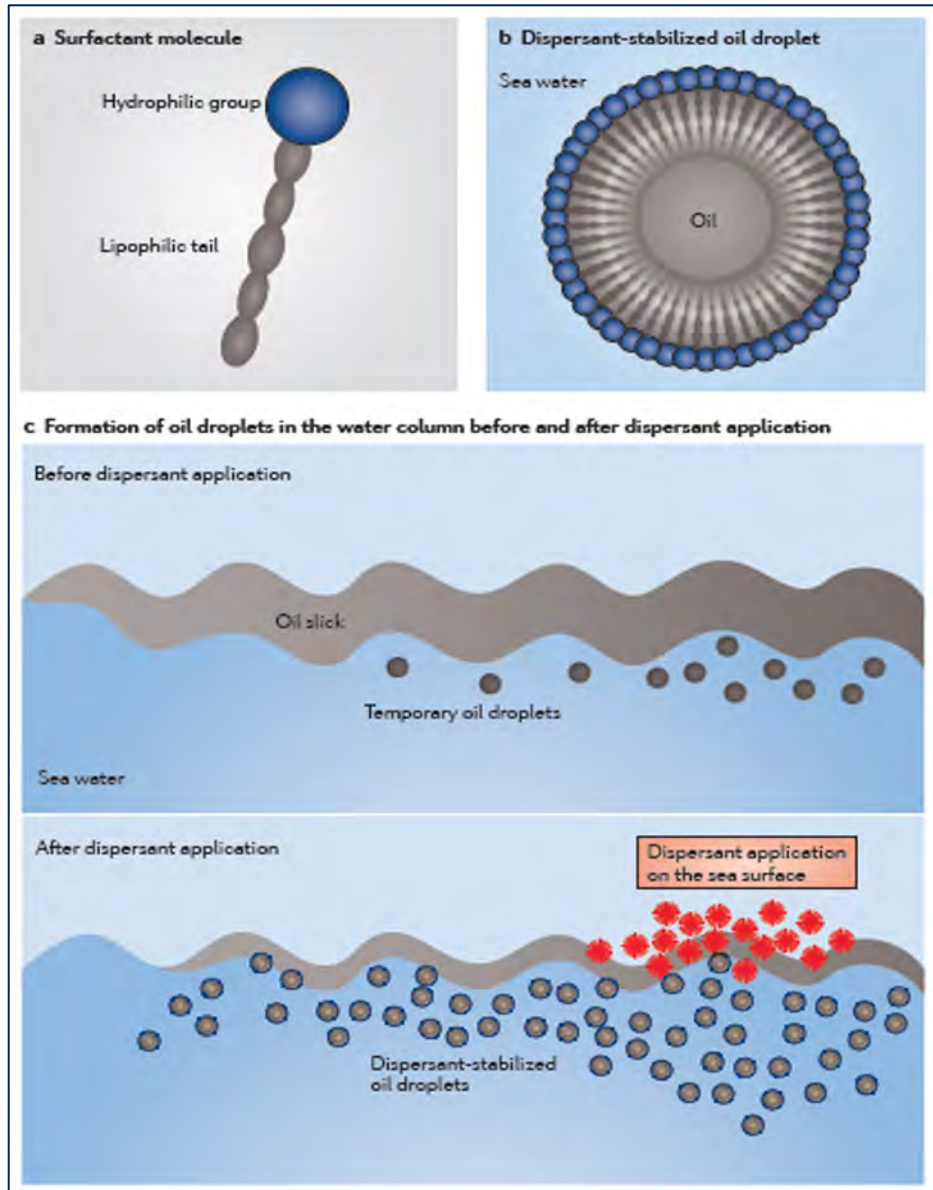
*Results*

4

*Conclusions*



# INTRODUCCIÓN



**Figure 1.** Action Mechanism and efficiency of biosurfactants in a petroleum hydrocarbons-contaminated seawater medium.

Source: Using dispersants after oil spills: impacts on the composition and activity of microbial communities. 2015

## Sample Preparation and Bioremediation Study

**Table 1. Properties of the crude oil VASCONIA**

<b>API gravity</b>	24.27
<b>Density (kg m<sup>-3</sup>) @ 15°C</b>	907.9
<b>Sulfur Concentration(wt.%)</b>	0.833
<b>Reid Vapour Pressure (kPa)</b>	21.99
<b>Flash Point(°C)</b>	0
<b>Número de ácido total (mg KOH/g)</b>	0.21
<b>Viscosity @ 40°C(cSt)</b>	22
Source:: Ecopetrol S.A.	

**Table 2. Physicochemical Properties of the Seawater used.**

Parameters	Unit	Value
<b>pH</b>	pH units	8.58
<b>Turbidity</b>	NTU	0
<b>COD</b>	mgO <sub>2</sub> L <sup>-1</sup>	98
<b>DO</b>	mgO <sub>2</sub> L <sup>-1</sup>	7.70
<b>Salinity</b>	g of salts kg <sup>-1</sup> of water	25.8
<b>Chlorides</b>	mg Cl <sup>-</sup> L <sup>-1</sup>	15900
<b>Ortophosphates</b>	mg P.PO <sub>4</sub> L <sup>-1</sup>	0.102
<b>Total Microbial Count</b>	CFU /100mL	ND*

# MATERIALS AND METHODS



**Table 3. Samples used and their components**

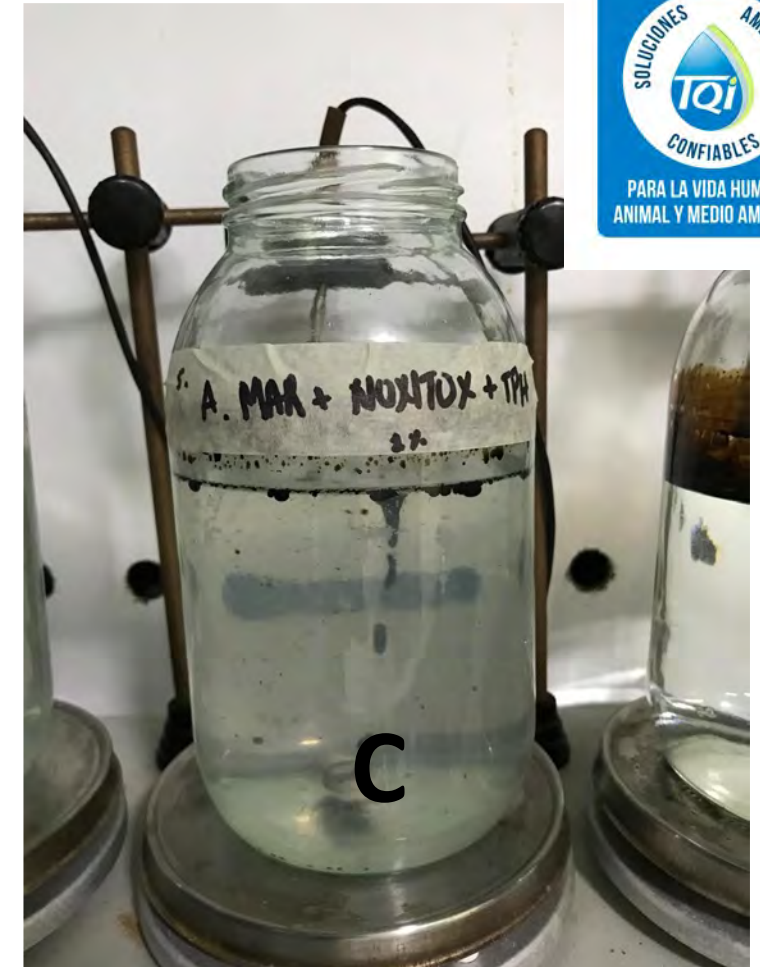
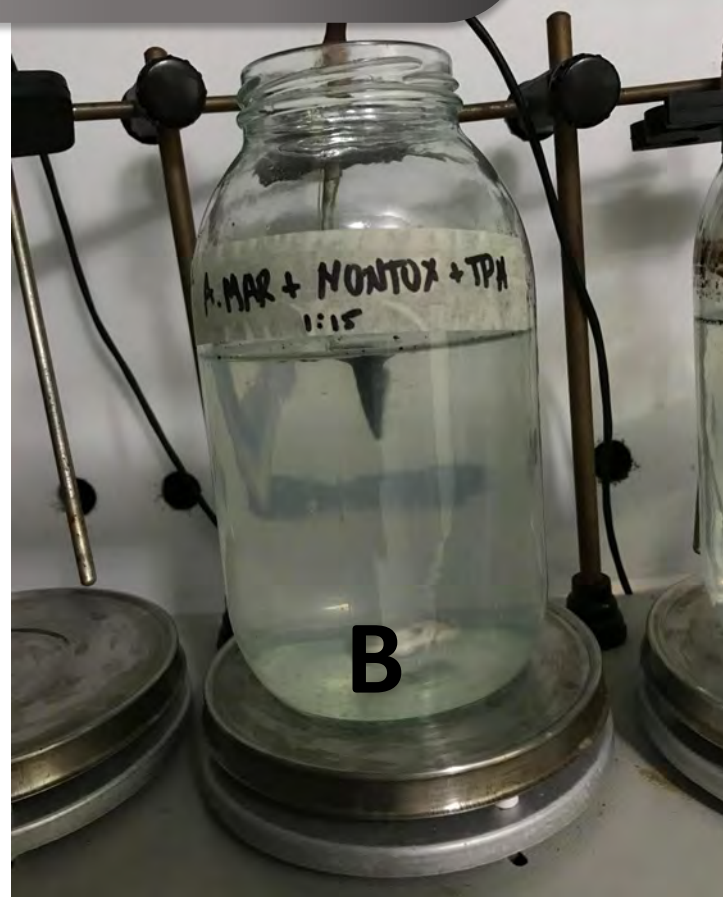
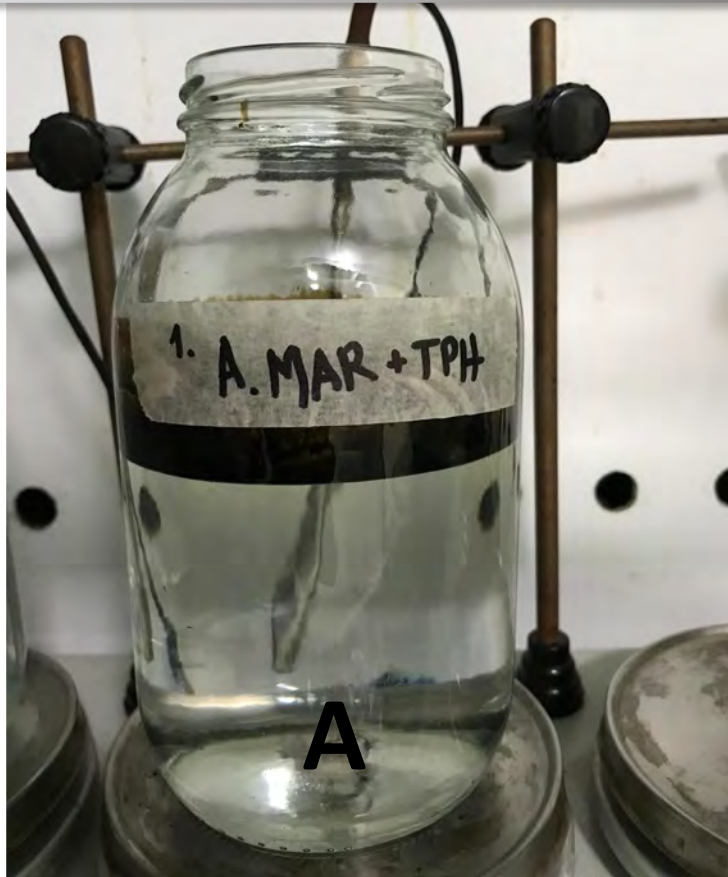
Sample	Components
A (Control)	Crude oil and seawater only ( $640 \text{ mg L}^{-1}$ )
B	Seawater, bio-catalytic agent solution ( $2167.39 \text{ mg L}^{-1}$ ), and crude oil ( $640 \text{ mg L}^{-1}$ )
C	Seawater, bio-catalytic agent solution ( $10000 \text{ mg L}^{-1}$ ) and crude oil ( $640 \text{ mg L}^{-1}$ ).
D (Control)	Seawater, bio-catalytic agent solution ( $2167.39 \text{ mg L}^{-1}$ ) and aeration system.
E	Seawater, bio-catalytic agent solution ( $2167.39 \text{ mg L}^{-1}$ ), crude oil ( $640 \text{ mg L}^{-1}$ ) and aeration system.
F	Seawater, bio-catalytic agent solution ( $10000 \text{ mg L}^{-1}$ ), crude oil ( $640 \text{ mg L}^{-1}$ ) and aeration system.

700 mL de of seawater, 0.5 mL of crude oil and 1.3 mL of bio-catalytic agent (0.216% v/v).



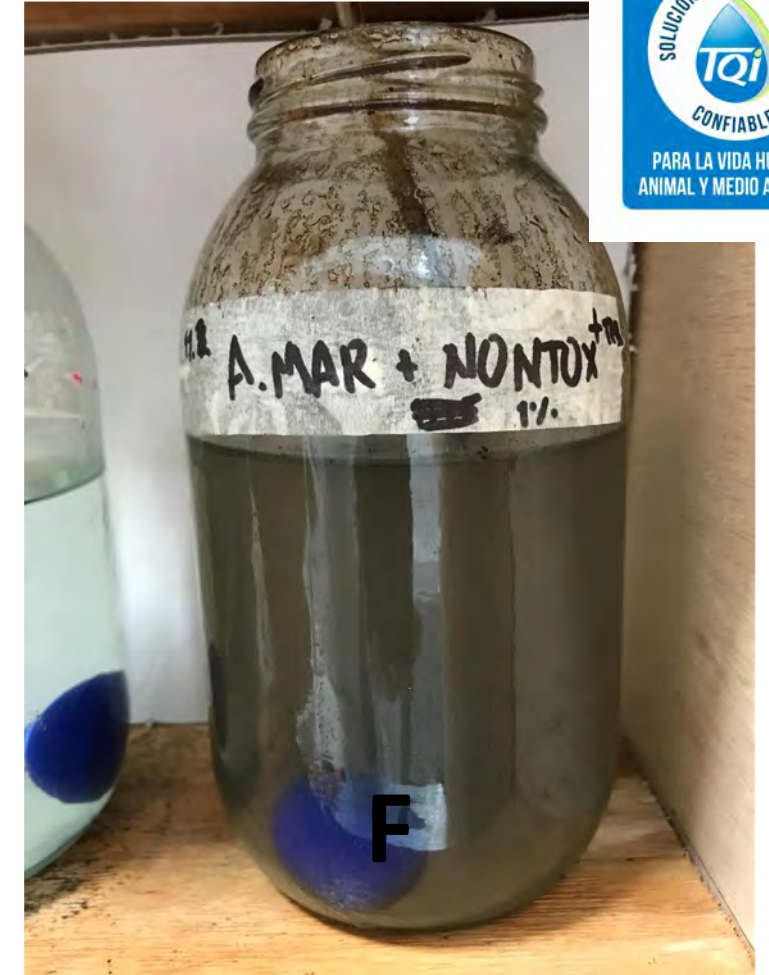
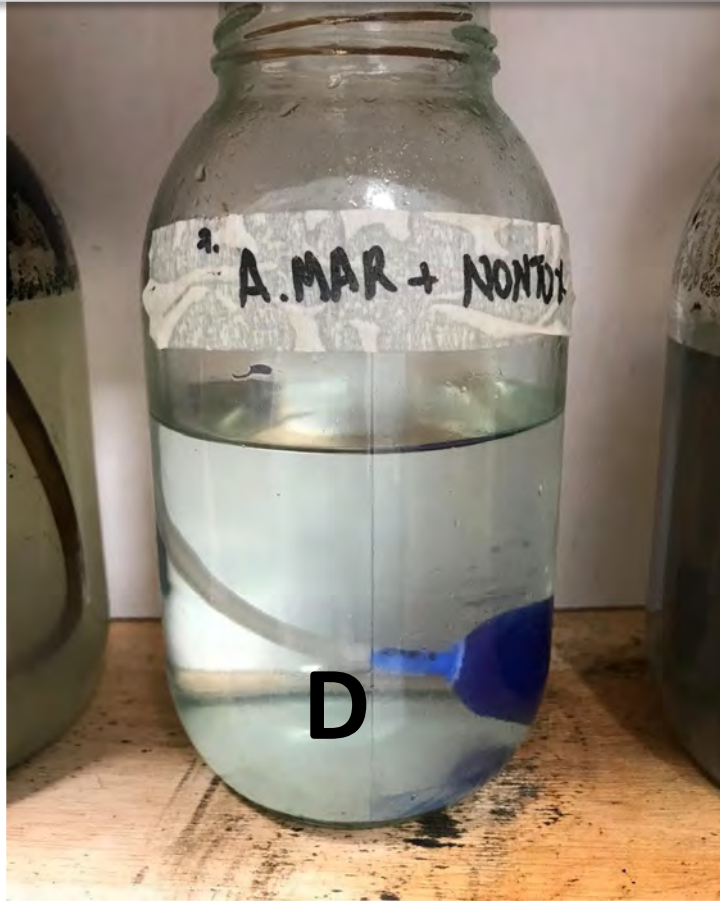


# MATERIALS AND METHODS



The samples labeled A, B, and C were subjected to an agitation system in a magnetic stirrer.

# MATERIALS AND METHODS



All the experimental set-up vessels were stored at 25 °C and average relative humidity of 64.5%.

The samples labeled D, E and F were agitated uninterruptedly for aeration and mixing to increase contact between the indigenous microbial consortium, nutrients, and contaminated water.

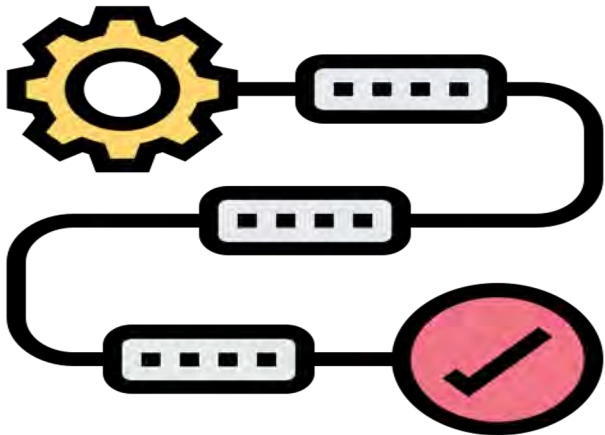


**Drop Collapse Test:** A clean flat surface was used to carry out the experiment, and the holes in there were filled with 5  $\mu\text{L}$  of vegetable oil and 5  $\mu\text{L}$  of bio-catalyst solution were added to the oil surface.

## Oil Spreading Assay:

5 mL of distilled water were poured into a 15 cm diameter Petri dish, followed by the addition of 100  $\mu\text{L}$  of Bazu oil to the surface of the water to form a thin layer of oil.

About 10  $\mu\text{L}$  of the bio-catalytic agent solution was added to the center of the formed oil layer, and the diameter of the cleaned area was measured.





# MATERIALS AND METHODS

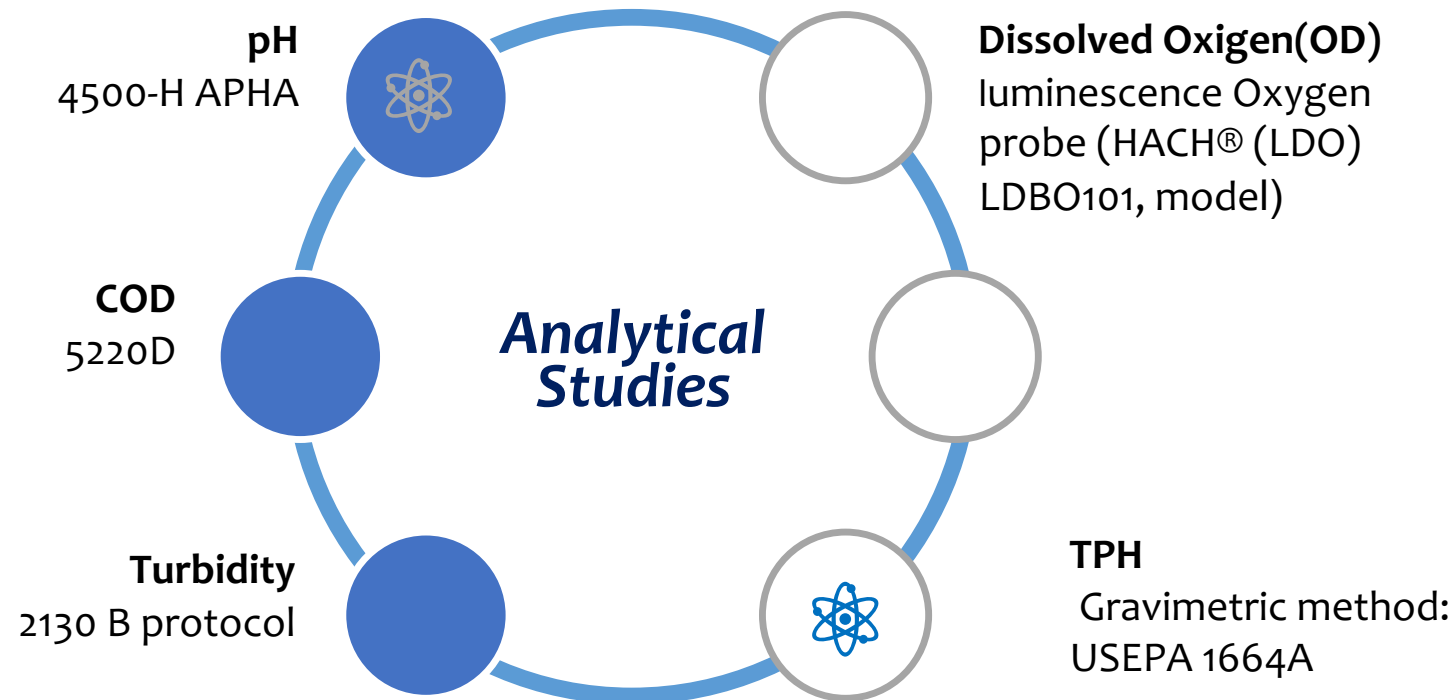


**Emulsification Test:** The test was realized by mixing 2 mL of kerosene with an equal volume of a biocatalytic agent, which was previously stirred in a vortex type agitator for 2 min and left to stand for twenty-four (24) hours.

## **Critical Micelle Concentration(CMC):**

This test required the preparation of different dilutions of a biocatalytic agent in distilled water. The changes in surface tension were measured in a digital tensiometer at a temperature of 25 °C, and distilled water of 96% purity was used as standard.

# MATERIALS AND METHODS

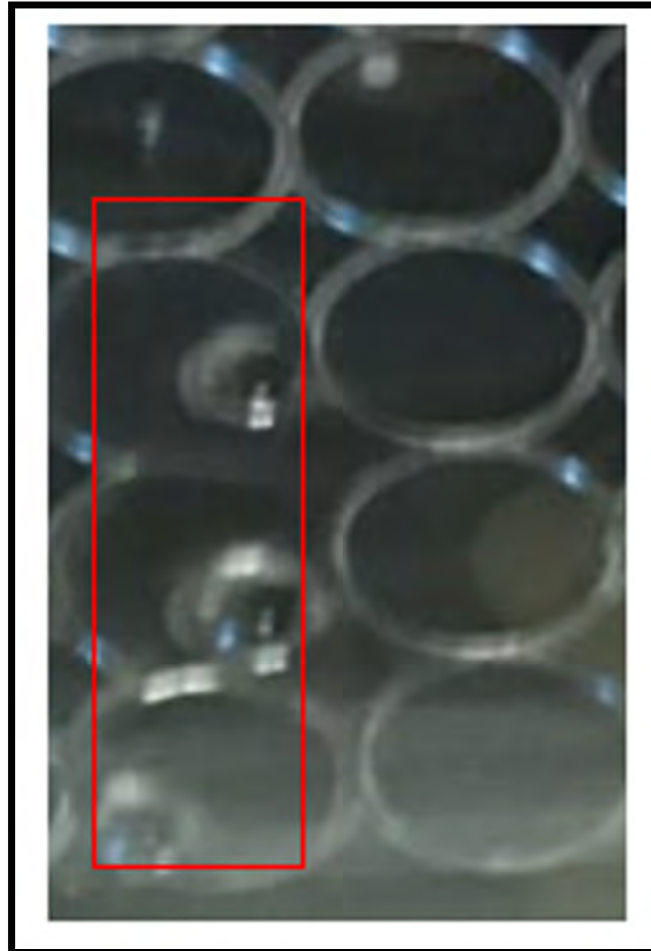


All these physico-chemical properties were analyzed according to the *Standard Methods for examination of water and Wastewater*



# RESULTS

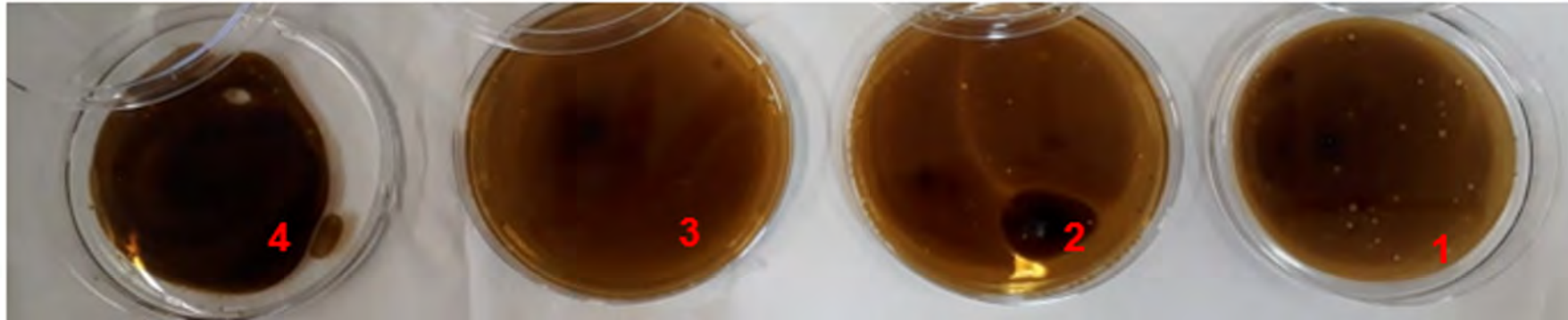
## Qualitative Characterization



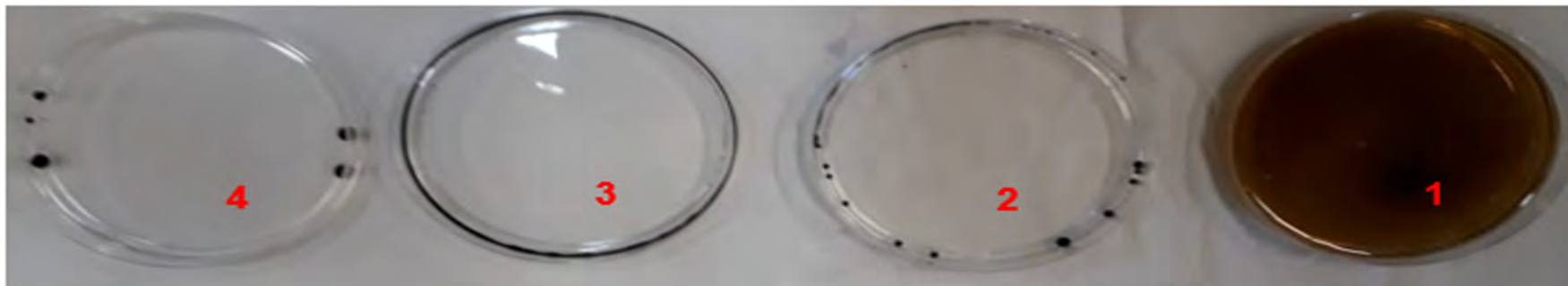
**Figure 2.** Drop Collapse Test Result. The drop did not retain its shape after 1 min of inspection.

# RESULTS

## Qualitative Characterization



**Figure 3.** Oil-Spreading assay before bio-catalytic solution application.

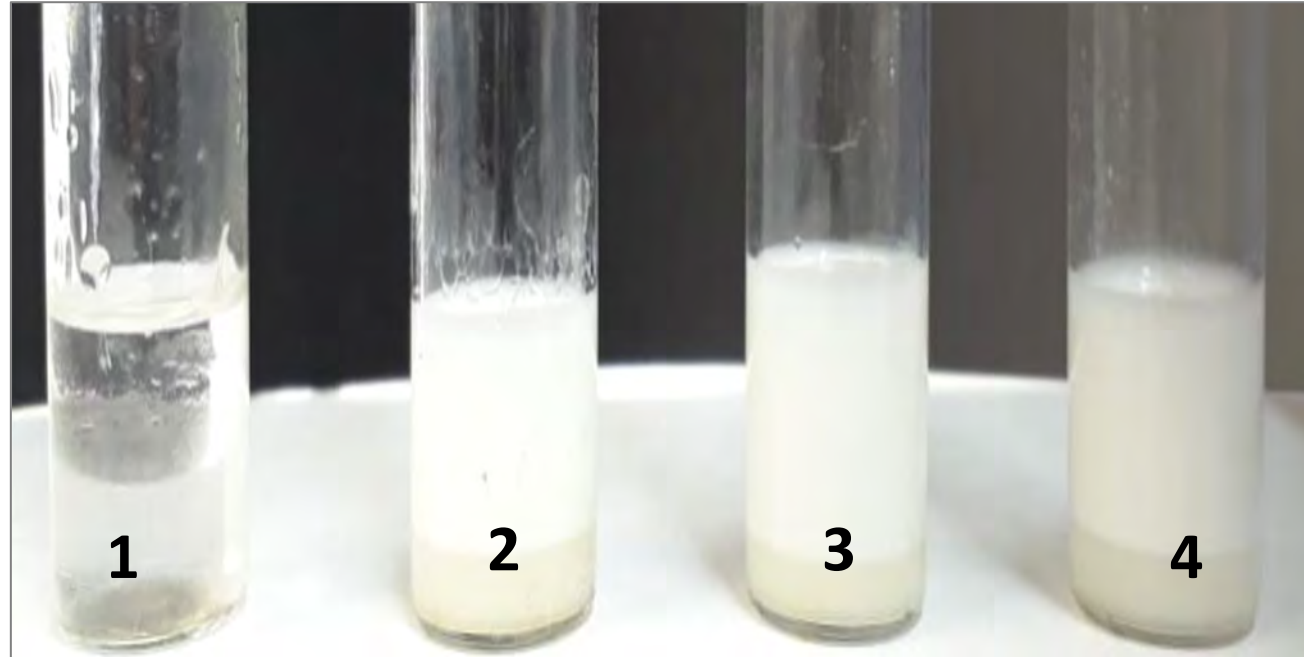


**Figure 4.** Oil-Spreading assay after bio-catalytic solution application.

(1) sample with crude oil and distilled water (negative control) ;(2), (3) y (4). sample with crude oil, distilled water and bio-catalytic solution.



# Qualitative Assays



**Figure 5.** Test-tube used to measure the emulsification index (E24), (1). The first test-tube is the control sample whose composition is only 2 mL kerosene. (2), (3), (4). These test-tubes contain a mixture of 2 mL kerosene with 2 mL of bio-catalytic agent solution previously stirred in a vortex type agitator for 2 min.

Average emulsification index: **74.47 %  $\pm$  5.55**

# Qualitative Assays



**Table 4.** Surface Tension values measured at a given concentration of biocatalytic agent.

Bio-catalytic Agent Concentration (mg L <sup>-1</sup> )	Surface Tension (mN m <sup>-1</sup> )
540	30.0 ± 0.08
430	30.2 ± 0.09
140	29.8 ± 0.16
70	29.4 ± 0.25
60	29.6 ± 0.11
50	29.8 ± 0.41
40	29.0 ± 0.10
30	29.2 ± 0.64
20	29.8 ± 0.70
10	29.2 ± 1.05
5	29.4 ± 1.23
0 (Distilled water without bio-catalyst nor oil)	73.1 ± 0.95

Values below 40 mN/m and the concentration varies between 1 and 200 mg/L

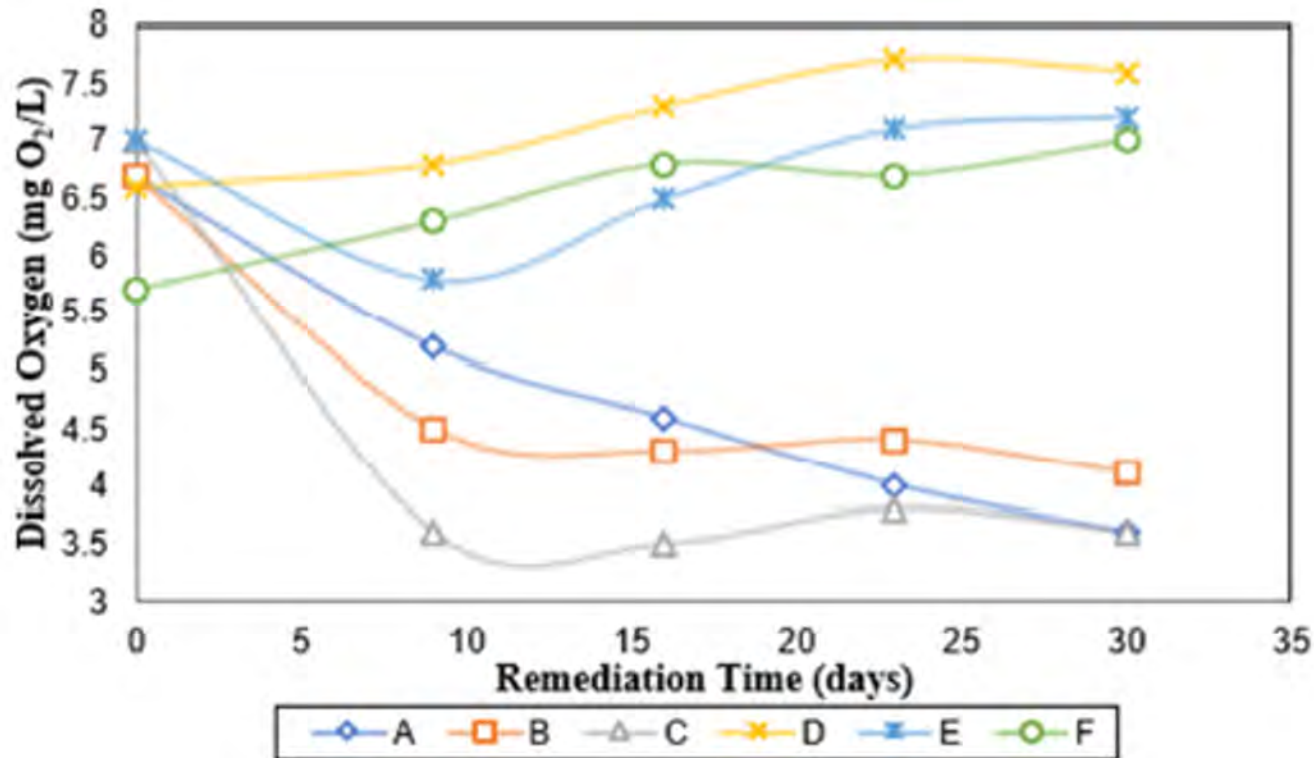


# Results

## Physicochemical Properties Analysis



Figure 6. Variation of dissolved oxygen content with remediation time.



It was appreciated that the DO increased with remediation time for the samples labeled D, E, and F, where the cracking of hydrocarbons was effective.

**Bio-catalytic agent capacity to improve the Oxygen transfer and speed up the degradation of organic matter.**

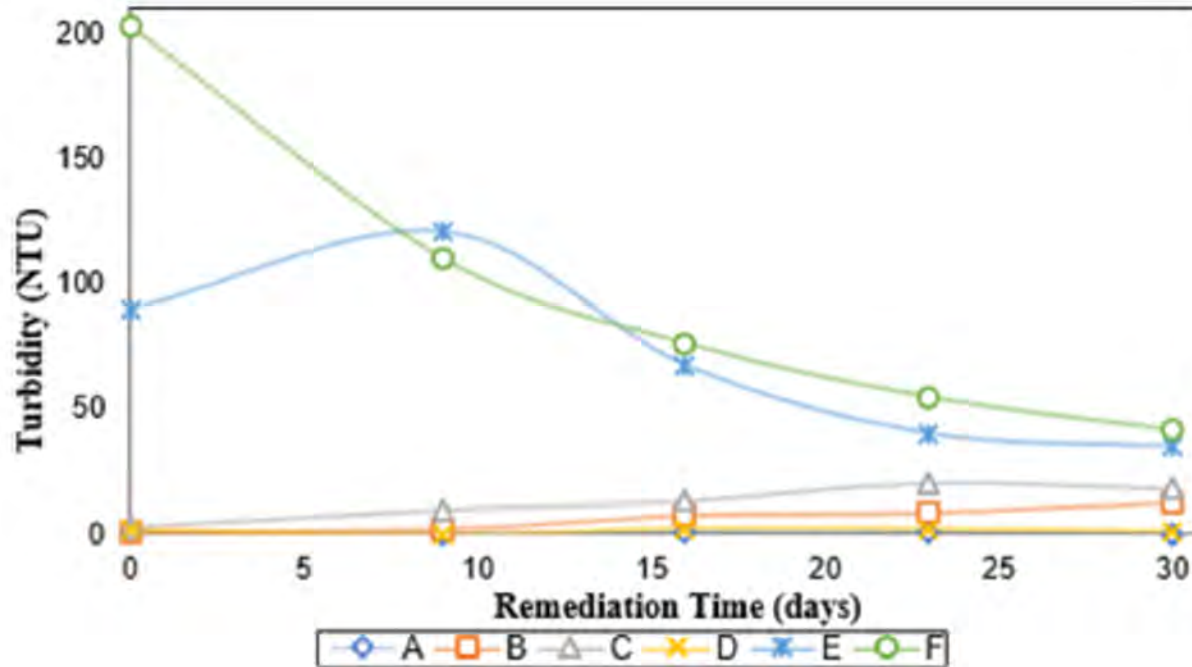
(A) Seawater with oil (640 mg L<sup>-1</sup>); (B) Seawater with oil (640 mg L<sup>-1</sup>) and biocatalytic agent solution (2167 mg L<sup>-1</sup>); (C) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (10000 mg L<sup>-1</sup>); (D) Seawater with bio-catalytic agent solution (2167.39 mg L<sup>-1</sup>) and aeration system; (E) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (2167 mg L<sup>-1</sup>), and aeration system; (F) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (10000 mg L<sup>-1</sup>) and aeration system.

# Results

## Physicochemical Properties Analysis



Figure 7. Variation of turbidity with remediation time



The samples labeled E and F achieved turbidity reduction values of  $61.357\% \pm 0.053$  and  $79.623\% \pm 0.053$ , respectively.

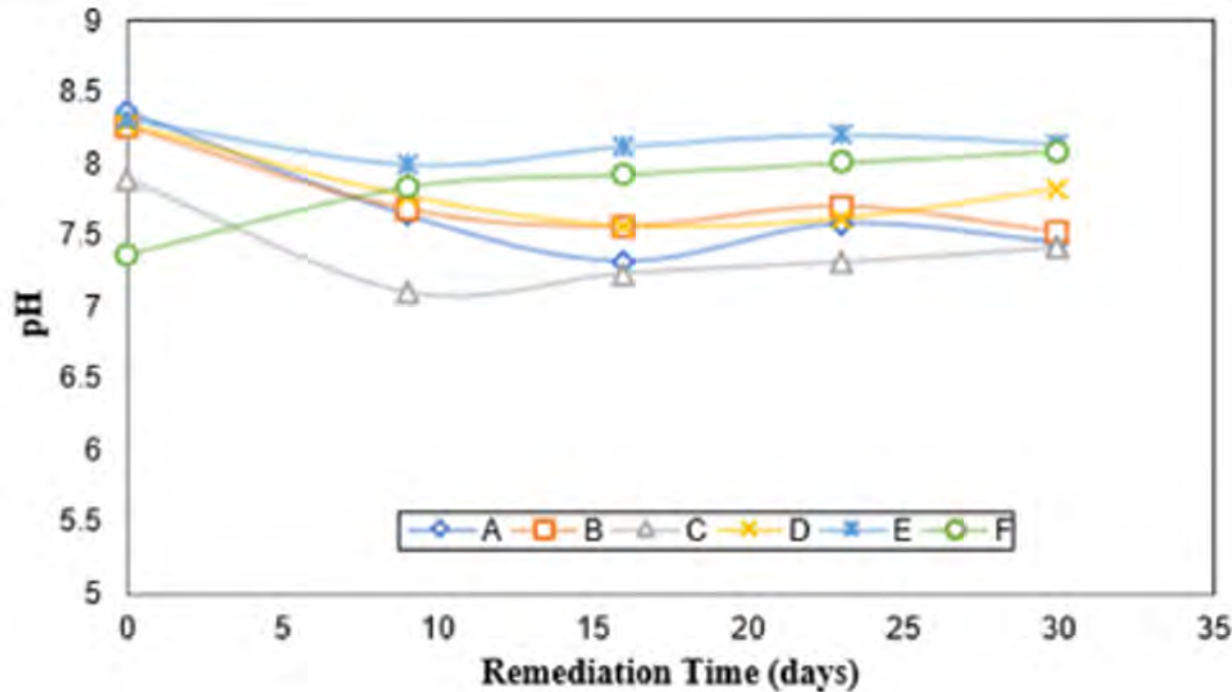
(A) Seawater with oil(640 mg L<sup>-1</sup>); (B) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution(2167 mg L<sup>-1</sup>); (C) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (10000 mg L<sup>-1</sup>); (D) Seawater with bio-catalytic agent solution(2167.39 mg L<sup>-1</sup>) and aeration system; (E) Seawater with oil (640 mg L<sup>-1</sup>), biocatalytic agent solution (2167 mg L<sup>-1</sup>), and aeration system; (F) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (10000 mg L<sup>-1</sup>) and aeration system.

# Results

## Physicochemical Properties Analysis



Figure 8. Variation of pH with remediation time



The pH of the sample labeled F exhibited a continuous rising in pH values, indicating that pollutant (petroleum hydrocarbons) in the water was decomposed to compounds that are more basic and less toxic

((A) Seawater with oil(640 mg L<sup>-1</sup>); (B) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution(2167 mg L<sup>-1</sup>); (C) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (10000 mg L<sup>-1</sup>); (D) Seawater with bio-catalytic agent solution(2167.39 mg L<sup>-1</sup>) and aeration system; (E) Seawater with oil (640 mg L<sup>-1</sup>), biocatalytic agent solution (2167 mg L<sup>-1</sup>), and aeration system; (F) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (10000 mg L<sup>-1</sup>) and aeration system.

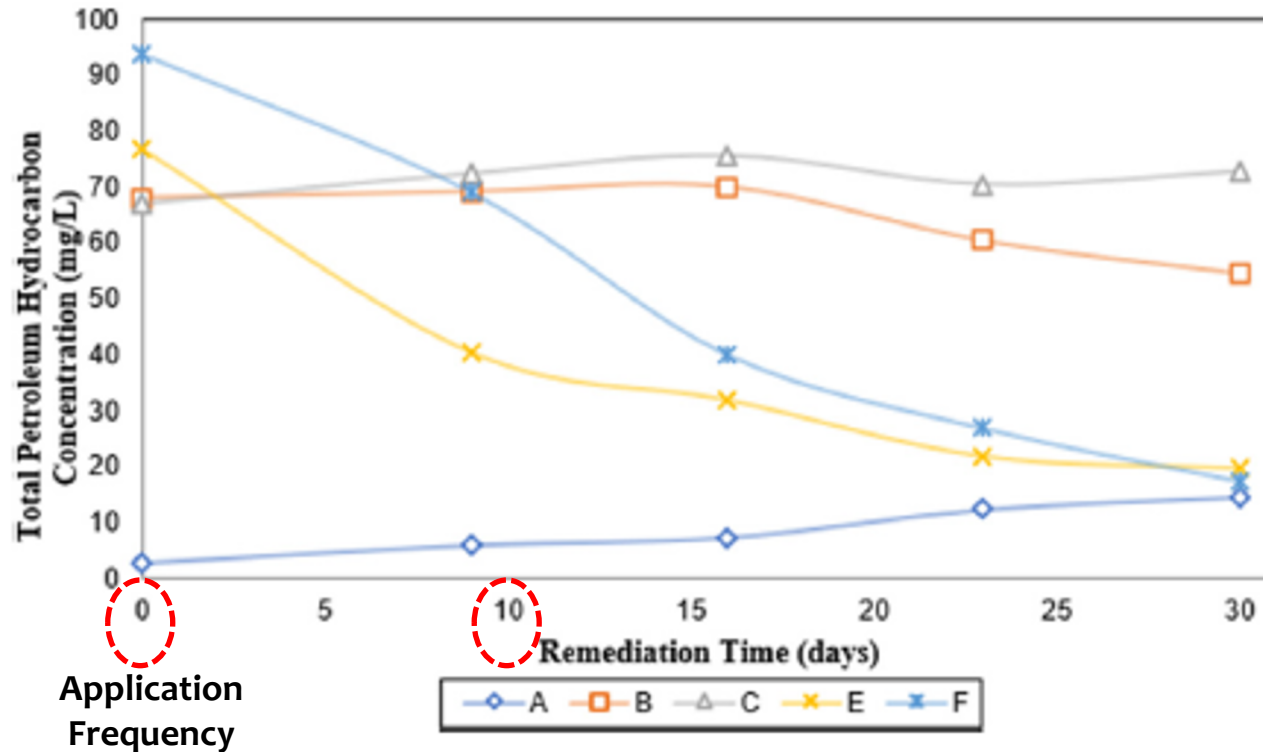


# Results

## Physicochemical Properties Analysis



Figure 9. Variation of total petroleum hydrocarbon concentration with remediation time.



The presence of a bio-catalyst solution and an aeration system generates the formation of micro-bubble. These micro-bubbles are the result of aggregates of surfactant molecules with a loose molecular packing, which provokes a more favorable Oxygen mass transfer into an aqueous medium.

As we expected, the stimulated sample labeled F reached the highest %RTPH with a rough value of  $81.537 \pm 0.015$ , followed by  $74.446 \pm 0.015$  corresponding to sample labeled E.

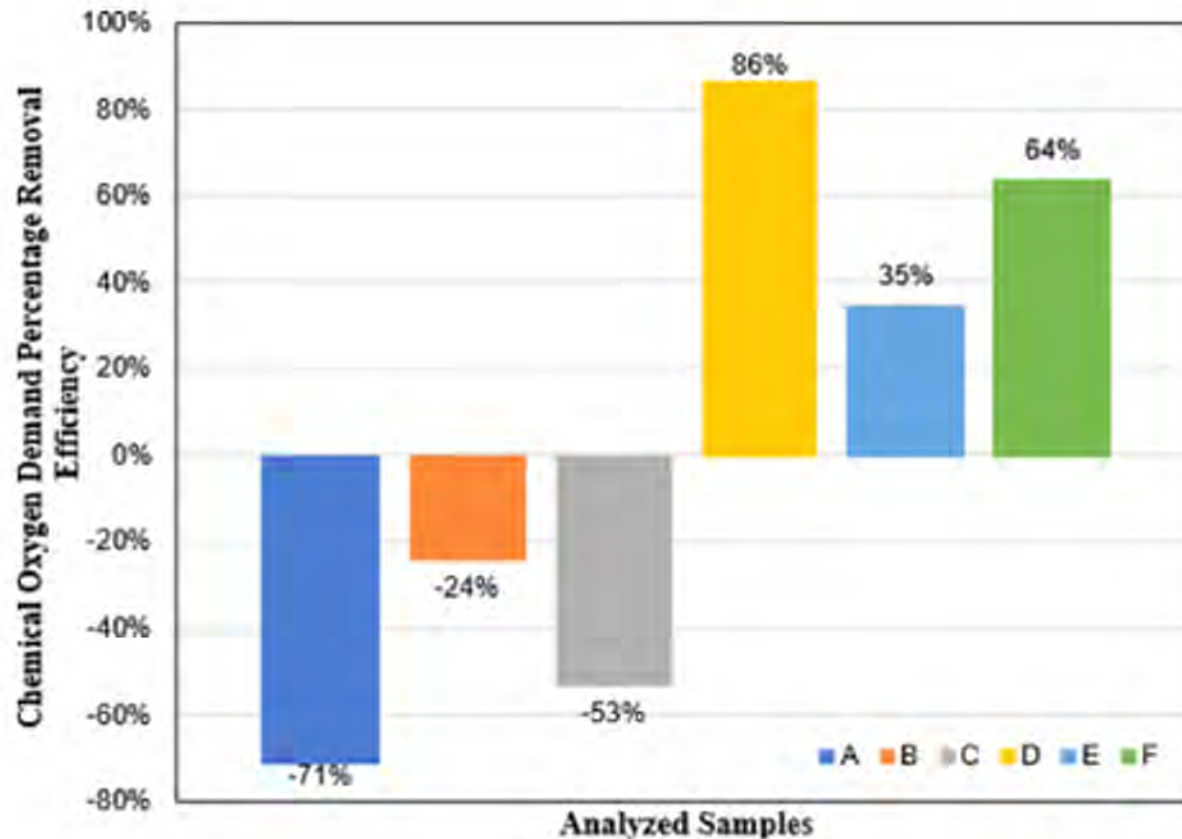
(A) Seawater with oil (640 mg L<sup>-1</sup>); (B) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (2167 mg L<sup>-1</sup>); (C) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (10000 mg L<sup>-1</sup>); (E) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (2167 mg L<sup>-1</sup>), and aeration system; (F) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (10000 mg L<sup>-1</sup>) and aeration system.

# Resultados

## Physicochemical Properties Analysis



Figure 10. Percentage of Chemical Oxygen Demand removal.



The highest %RCOD was  $64.539 \pm 0.125$ , and it was attained by sample labeled F, which had the most concentrated bio-catalyst solution, followed by sample labeled E which had a removal efficiency of  $35.325\% \pm 0.125$  for COD at the end of the remediation time.

((A) Seawater with oil (640 mg L<sup>-1</sup>); (B) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (2167 mg L<sup>-1</sup>); (C) Seawater with oil (640 mg L<sup>-1</sup>) and bio-catalytic agent solution (10000 mg L<sup>-1</sup>); (D) Seawater with bio-catalytic agent solution (2167.39 mg L<sup>-1</sup>) and aeration system; (E) Seawater with oil (640 mg L<sup>-1</sup>), biocatalytic agent solution (2167 mg L<sup>-1</sup>), and aeration system; (F) Seawater with oil (640 mg L<sup>-1</sup>), bio-catalytic agent solution (10000 mg L<sup>-1</sup>) and aeration system.

# Remarks



The analyzed bio-catalyst showed positive outcomes for the drop-collapse test and oil-spreading assay.

The measurement of emulsification activity (E24) and Critical Micelle Concentration(CMC) displayed values of 74.47% and 40 mg/L, respectively.

The highest TPH removal efficiency was reached by the sample labeled F (81.537%), which contained 640 mg/L of petroleum and 10000 mg/L of biocatalyst solution.



# Remarks



Agitated and aerated systems have an essential effect on the bioremediation process. TPH removal efficiencies for aerated samples were in a range of 70%–82%, while, those not subjected to an agitation system achieved only a near value of 20%.

Catabolic cooperation between groups of microorganisms is important during the bioremediation process, because sometimes the complete petroleum hydrocarbons degradation by an only microorganism is not possible.

# REFERENCES

- ABBASIAN, Firouz, LOCKINGTON, Robin, MALLAVARAPU, Megharaj, y NAIDU, Ravi. A Comprehensive Review of Aliphatic Hydrocarbon Biodegradation by Bacteria. *Applied Biochemistry and Biotechnology*. 2015,176. pp.670–699.
- ADAMS, Randy; ZAVALA-CRUZ, Joel y MORALES-GARCÍA, Fernando. Concentración residual de hidrocarburos en el suelo del trópico II: Afectación a la fertilidad y su recuperación. *Asociación interciencia*.2008,7.pp.483-489.
- APARNA, Avasarala; SRINIKETHAN, Govindan y HEDGE, Simon. Effect of addition of biosurfactant produced by *Pseudomonas* sps. on biodegradation of crude oil. En: *Second International Proceedings of Chemical, Biological & Environmental Engineering*. Memorias.Singapore: 2011. pp. 71-75.
- . Production and characterization of biosurfactant produced by a novel *Pseudomonas* sp. 2B. *Colloids Surfaces B: Biointerfaces*.2012.95. pp.23-29.
- ARA, Hosna, RAHAMAN, Mohamed, ISLAM, Mohamed y MALLICK,Albert. A laboratory approach for determining effects of potential oil spillage on water quality of Sundarbans mangrove forest, Bangladesh. *Journal of Life and Earth Science*. 2009.3.pp.23-28.
- ASIMEA, Aminanyanaba, y SAM-WOBO, Sammy. The impact of hydrocarbon waste from brass oil terminal on the phytoplankton and periphyton communities of lower brass river, Niger Delta, Nigeria. *Journal of Emerging Trends in Engineering and Applied Sciences*. 2011,2(5). pp.729-733.
- ATLAS, Marilyn y BARTHA,Richard. Biodegradation of petroleum in seawater at low temperatures. *Canadian Journal of Microbiology*. 1972.18. pp.1851-1855.

# REFERENCES

- BODOUR, Adria y MILER-MAIER, Raina. Application of a modified drop-collapse technique for surfactant quantitation and screening of biosurfactant-producing microorganisms. *Journal of Microbiological Methods*. 1998. pp. 273–280.
- BOETHLING, Robert y ALEXANDER, Martin. Effect of concentration of organic chemicals on their biodegradation by natural microbial communities. *Applied and Environmental Microbiology*. 1979.37. pp.1211- 1216.
- CALVO, Concepción, MANZANERA, Maximinio, SILVA-CASTRO, Gina, UAD, Imane. y GONZÁLEZ-LÓPEZ, Jesús. Application of bioemulsifiers in soil oil bioremediation processes. Future prospects. *Science of Total Environment*. 2009.407. pp.3634–3640.
- CAMPOS, Jenyffer. Microbial biosurfactants as additives for food industries. *Biotechnology Progress*. 2013, 29 (5). pp.1097–1108.
- CAO,Bin; NAGARAJAN, Karthiga y Loh, Kai-Chee. Biodegradation of aromatic compounds: current status and opportunities for biomolecular approaches. *Applied Microbiology and Biotechnology*.2009.85(2).pp. 207–228
- CHANDRA, Subhash, SHARMA, Richa, SINGH, Kriti, y SHARMA, Anima. Application of bioremediation technology in the environment contaminated with petroleum hydrocarbon. *Annals of Microbiology*. 2013.63.pp.417–431.
- CHIRON, Serge; SAUVAR, Emmanuel y JEANNOT, R. Determination of nonionic polyethoxylated surfactants in wastewater and sludge samples of sewage treatment plants by liquid chromatography-mass spectrometry *Analisis*.2000,28.pp.535.



# REFERENCES

GAFAROV, Anton, FILONOV, Anatoliy, PANOV, Andrey y BORONIN, Am. Change in the composition of a bacterial association degrading aromatic compounds during oil sludge detoxification in a continuous-flow microbial reactor. *Applied Biochemistry and Microbiology*. 2006, 42. pp. 160–165.

GAMBA, Karen y PEDRAZA, Angie. Evaluación de estrategias de biorremediación para el tratamiento de aguas residuales industriales contaminadas con aceites usados. *Ingeciencia*. 2018, 2(2). pp. 18-30.

GERVAJIO, Gregorio; WITHANA-GAMAJE, y Thushand y SIVAKUMAR, Mahesh. Fatty acids and coconut derivatives oils. En: BAILEY, Alton. *Bailey's Industrial Oil and Fat Products*. Seventh Edition. Edited. Fereidoon Shahidi, 2020.

GORBANYOV, Vladimir. Green technology and sustainable development of environment. *Renewable Res. J.* 2015, 3(1). p. 244–249.

GUWAHATI lit. Contaminated site remediation. National Programme on Technology Enhanced Learning. 2012.

HÖHENER, Patrick, DUWIG, Celine, PASTERIS, Gabrieli, DAKHEL, Nathalie, KAUFMANN, Karin y HARMS, Hauke. Biodegradation of petroleum hydrocarbon vapors: Laboratory studies on rates and kinetics in unsaturated alluvial sand. *Journal of Contaminant Hydrology*. 2003, 66. pp. 93 – 115

LEAHY, Joseph y COLWELL, Rita. Microbial degradation of hydrocarbons in the environment. *Microbiology and Molecular Biology Reviews*. 1990, 54(3). pp. 305-315.

# REFERENCES

- YOUSSEF, N. H., DUNCAN, K. E., NAGLE, D. P., SAVAGE, K. N., KNAPP, R. M., AND MCINERNEY, M. J. Comparison of methods to detect biosurfactant production by diverse microorganisms. *Journal of Microbiological Methods*. 2004,56. pp.339–347.
- ZAGHDEN, Hatem, KALLEL, Monem, LOUATI, Afifa, ELLEUCH, Bboubaker, OUDOT, Jean y SALIOT, Antoine Hydrocarbons in surface sediments from the Sfax coastal zone (Tunisia) Mediterranean Sea. *Marine Pollution Bulletin*. 2005.50(11).pp.1287–1294.
- ZEMBRZUSKA, J. Determination of Dodecanol and Short-Chained Ethoxylated Dodecanols by LC–MS/MS (with Electrospray Ionization) after their Derivatization (with Phenyl Isocyanate). *Journal Surfactant Detergent*. 2017,20. p.1421–1432.
- ZEMBRZUSKA, J., BUDNIK, I., AND LUKASZEWSKI, Z. Parallel pathways of ethoxylated alcohol biodegradation under aerobic conditions. *Science of The Total Environment*, 557-558, 612–619.
- ZHANG, Y y MILLER. Enhanced octadecane dispersion and biodegradation by a *Pseudomonas rhamnolipid* surfactant (biosurfactant). *Applied and Environmental Microbiology*. 1992,58. pp.3276–3282.

A large oil tanker ship is silhouetted against a bright orange sunset sky. The ship's complex structure, including cranes and masts, is visible. The water in the foreground is dark and textured. The text "Thanks for your attention" is overlaid in white, with "your" partially obscured by the ship's silhouette.

**Thanks for  
your attention**

**Questions?**





## Contact Information

Email Address: [gteran@est.uniatlantico.edu.co](mailto:gteran@est.uniatlantico.edu.co)

Mobile: (+966)0544502959  
(+57)3046477216

Email Address: [camila.Carvajalino@tqi.co](mailto:camila.Carvajalino@tqi.co)

Mobile: (+57)3015546337

Email Address: [luz.agamez@tqi.co](mailto:luz.agamez@tqi.co)

Mobile: (+57) 3213661525